

Magnetic vs. nonmagnetic polymorphs of RuBr_3 under pressure

Bin Shen,^{1,*} Victoria A. Ginga,^{2,*} Angel M. Arévalo-López,³ Gaston Garbarino,⁴ Ece Uykur,⁵ Marcos Gonçalves-Faria,⁵ Prashanta K. Mukharjee,¹ Philipp Gegenwart,¹ and Alexander A. Tsirlin^{2,†}

¹*Experimental Physics VI, Center for Electronic Correlations and Magnetism,
University of Augsburg, 86159 Augsburg, Germany*

²*Felix Bloch Institute for Solid-State Physics, University of Leipzig, 04103 Leipzig, Germany*

³*UMR-8181-UCCS-Unité de Catalyse et Chimie du Solide-Univ. Lille,
CNRS, Centrale Lille, ENSCL, Univ. Artois, F-59000 Lille, France*

⁴*European Synchrotron Radiation Facility, 38043 Grenoble, France*

⁵*Helmholtz Zentrum Dresden Rossendorf, Inst Ion Beam Phys & Mat Res, D-01328 Dresden, Germany*

Pressure evolution of the crystal structure and magnetism of the honeycomb α - RuBr_3 is studied using high-pressure x-ray diffraction, magnetometry, and density-functional band-structure calculations. Hydrostatic compression transforms antiferromagnetic α - RuBr_3 ($R\bar{3}$) into paramagnetic α' - RuBr_3 ($P\bar{1}$) where short Ru–Ru bonds cause magnetism collapse above 1.3 GPa at 0 K and 2.5 GPa at 295 K. Below this critical pressure, the Néel temperature of α - RuBr_3 increases with the slope of 1.8 K/GPa. Pressure tunes α - RuBr_3 away from the Kitaev limit, whereas increased third-neighbor in-plane coupling and interlayer coupling lead to a further stabilization of the collinear zigzag state. Both α - and α' - RuBr_3 are metastable at ambient pressure, but their transformation into the thermodynamically stable β -polymorph is kinetically hindered at room temperature.

I. INTRODUCTION

Experimental realization of the Kitaev model and its intriguing spin-liquid physics [1] requires honeycomb magnets with the d^5 or d^7 transition-metal ions [2, 3]. The choice of chemical compounds satisfying these criteria appears to be quite limited, especially in the case of Ru^{3+} ($4d^5$) that has been known to form only one honeycomb magnet, the widely studied α - RuCl_3 [4, 5]. Other ruthenium trihalides exist too, but they adopt chain-like structures and show mundane temperature-independent paramagnetic behavior [6]. A similar chain-like structure reported for the chloride is commonly identified as β - RuCl_3 in contrast to the α -polymorph with the honeycomb layers [7–9].

Relative stability of the α - and β -polymorphs of RuCl_3 is controlled by temperature. Whereas β - RuCl_3 is synthesized at 600 – 650 K, increasing the synthesis temperature above 700 K stabilizes the magnetic α -polymorph [10]. A somewhat similar transformation is also possible in the bromide. Its chain-like paramagnetic form, β - RuBr_3 , is the only polymorph that can be synthesized at ambient pressure [11]. However, the high-pressure high-temperature treatment of β - RuBr_3 leads to the honeycomb structure of α - RuBr_3 [12, 13] that shares many similarities with α - RuCl_3 , including its local magnetism of Ru^{3+} and collinear zigzag magnetic order at low temperatures [12, 14, 15]. External pressure and temperature can thus control the formation of magnetic vs. nonmagnetic polymorphs of the Ru^{3+} trihalides.

High-pressure treatment renders RuBr_3 magnetic. This evolution is remarkably different from α - RuCl_3 and honeycomb iridates that typically become nonmagnetic upon application of pressure [16–18], thus limiting the use of pressure as a tuning parameter. The magnetism collapse in all these Kitaev materials is caused by the formation of short metal-metal bonds that break regular

honeycombs into nonmagnetic dimers [19–22]. The same mechanism is in fact responsible for the temperature-independent paramagnetic behavior of β - RuBr_3 where metal-metal bonds (dimers) are formed within the ruthenium chains [11, 13, 23, 24], see Fig. 1. This collapsed magnetic state is suppressed when β - RuBr_3 transforms into the α -polymorph upon application of pressure. It raises an interesting question whether α - RuBr_3 may be more robust against pressure-induced magnetism collapse than α - RuCl_3 , thus offering a broader pressure window for tuning Kitaev magnetism in the honeycomb planes. The larger unit-cell volume of α - RuBr_3 compared to α - RuCl_3 should also facilitate the stability of this compound against the pressure-induced structural dimerization.

In the following, we investigate α - RuBr_3 under hydrostatic pressure and report its structural evolution as well as magnetic behavior. We show that α - RuBr_3 follows the same scenario of pressure-induced structural dimerization as the chloride, but with the magnetism collapse happening at higher pressures. This similar behavior contrasts with the different pressure evolution of T_N in the chloride and bromide. We also assess thermodynamic stability of the RuBr_3 polymorphs and reveal the metastable nature of α - RuBr_3 at ambient pressure.

II. METHODS

Polycrystalline samples of α - RuBr_3 were prepared by annealing the commercial β - RuBr_3 powder (Alfa Aesar) at 6 GPa and 900 °C in a Walker-type multianvil press followed by cooling the sample to room temperature before releasing the pressure. The recovered powder contained the $R\bar{3}$ phase (α - RuBr_3) with 1.6 wt.% of the RuO_2 impurity according to the Rietveld refinement of x-ray diffraction (XRD) data. No traces of the β -

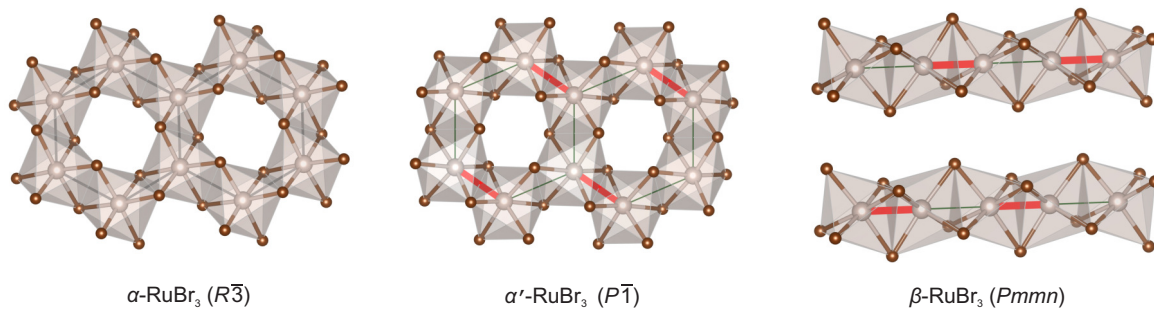


FIG. 1. Crystal structures of the RuBr_3 polymorphs. The red lines show the Ru–Ru dimers.

polymorph were detected.

High-pressure XRD data were collected at room temperature at the ID27 beamline of the European Synchrotron Radiation Facility (ESRF, Grenoble, France) using the wavelength of 0.3738 Å and EIGER2 X CdTe 9M detector. Powder samples of $\alpha\text{-RuBr}_3$ and $\beta\text{-RuBr}_3$, respectively, were placed into a stainless-steel gasket mounted inside a diamond anvil cell filled with helium gas as pressure-transmitting medium. Pressure was measured using the fluorescence line of a ruby sphere placed into the cell next to the sample. Two-dimensional images were integrated using the *Dioplas* software [25]. *Jana2006* [26] was used for structure refinement. The high-pressure XRD data were collected up to 17 GPa for $\alpha\text{-RuBr}_3$ and up to 12 GPa for $\beta\text{-RuBr}_3$.

Magnetization under pressure was measured similar to Ref. [17]. Pressure was calibrated by measuring the superconducting transition of Pb. Daphne oil 7373 was used as pressure-transmitting medium.

Density-functional (DFT) band-structure calculations were performed in the *VASP* code [27, 28] using the Perdew-Burke-Ernzerhof (PBE) exchange-correlation potential [29] with Grimme’s D3 dispersion correction [30] for weak van der Waals bonding, which is expected in the α - and β -polymorphs between the layers and chains, respectively. Crystal structures of different polymorphs were optimized at several constant volumes to obtain the equation of state and calculate enthalpies as a function of pressure. Additionally, *FPL0* [31] was used to calculate PBE band structures of $\alpha\text{-RuBr}_3$ at different pressures using experimental lattice parameters. Atomic positions were optimized at each pressure prior to constructing the tight-binding models via Wannier projections [32] and estimating exchange couplings using the superexchange model developed in Refs. [33, 34]. First Brillouin zone was sampled by a fine mesh with up to 700 k -points in the symmetry-irreducible part.

III. RESULTS

A. Crystal structure

Our $\alpha\text{-RuBr}_3$ sample shows the $R\bar{3}$ crystal structure at ambient pressure in agreement with the previous report [12]. This rhombohedral structure remains stable up to 2.5 GPa at room temperature. At higher pressures, an abrupt change in the XRD patterns (Fig. 2c) indicates a phase transition toward the high-pressure polymorph with lower symmetry. This $\alpha'\text{-RuBr}_3$ polymorph is triclinic ($P\bar{1}$), similar to the high-pressure dimerized phase of RuCl_3 [16]. No further pressure-induced transformations were observed up to 17 GPa, the highest pressure of our experiment [35]. The α – α' transformation is fully reversible. Upon releasing pressure, the α -polymorph was recovered, albeit with the increased diffuse scattering indicative of the higher number of stacking faults that develop in the weakly bonded layered structure upon pressure cycling [35].

The pressure-induced structural phase transition is accompanied by the volume drop of about 3% (Fig. 2a), which is comparable to the 2.3% volume drop in $\beta\text{-Li}_2\text{IrO}_3$ [22] and 1.9% in $\alpha\text{-Li}_2\text{IrO}_3$ [19] upon their pressure-induced phase transitions. The Ru–Ru distances extracted from the structure refinements show that RuBr_3 also undergoes a structural dimerization, resulting in the magnetism collapse (see Sec. III C). Three equivalent Ru–Ru distances of about 3.6 Å in $\alpha\text{-RuBr}_3$ split into one short and two long, almost equal distances in the high-pressure phase (Fig. 2b). The shorter distance is about 2.9 Å right above the transition and still exceeds the Ru–Ru distance of 2.71 – 2.73 Å in $\beta\text{-RuBr}_3$ [11, 13], but this dimer distance in $\alpha'\text{-RuBr}_3$ rapidly shrinks and goes below 2.7 Å above 5 GPa. The rapid reduction in the Ru–Ru distances between 2.5 and 5 GPa may be caused by the large size of bromine that increases the average Ru–Ru distance in $\alpha\text{-RuBr}_3$ compared to the chloride and makes it more difficult to form the dimers. Higher pressure is thus required to complete the dimer formation.

A comparative pressure-dependent study of $\beta\text{-RuBr}_3$ showed that this polymorph does not transform into the layered (α - or α' -) structure up to at least 12 GPa at room

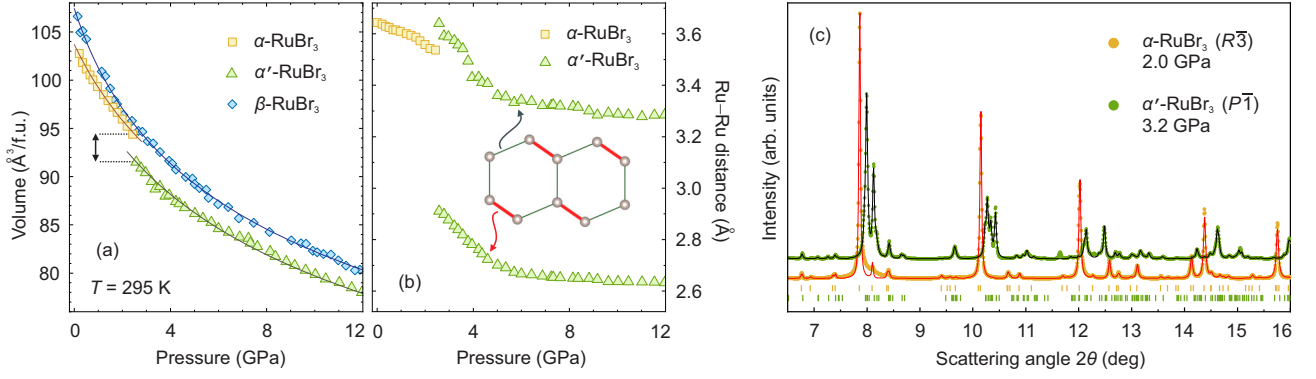


FIG. 2. (a) Pressure dependence of the unit-cell volume measured by XRD. The solid lines are the fits with Eq. (2). The dotted lines highlight the volume drop upon the $\alpha - \alpha'$ transition. (b) Pressure dependence of the Ru-Ru distances. The two longer distances in the α' -polymorph have been averaged for clarity. (c) XRD patterns of α - and α' -RuBr₃ (offset for clarity). The lines show the Le Bail fits.

temperature. It undergoes a steady compression [35] with a similar pressure dependence as in α' -RuBr₃, albeit with the larger unit-cell volume (Fig. 2a). Below we show that the $\beta - \alpha'$ transition should be thermodynamically favored above 5.5 GPa, and indeed a transformation into the layered polymorph occurs upon the high-pressure high-temperature treatment. The persistence of β -RuBr₃ up to much higher pressures at room temperature indicates that this transformation is kinetically hindered.

B. Thermodynamics

To assess thermodynamic stability of the different polymorphs, we calculated their total energies at several fixed volumes (Fig. 3a) and fitted these energies to the Murnaghan equation of state,

$$E(V) = E_0 + B_0 V_0 \left[\frac{1}{B'_0(B'_0 - 1)} \left(\frac{V}{V_0} \right)^{1-B'_0} + \frac{1}{B'_0} \frac{V}{V_0} - \frac{1}{B'_0 - 1} \right] \quad (1)$$

where B_0 is the bulk modulus at ambient pressure, B'_0 shows linear pressure dependence of the bulk modulus, while E_0 and V_0 are the equilibrium energy and volume, respectively. These parameters listed in Table I show a good agreement with the experimental values obtained by fitting the $V(P)$ curves to

$$V(P) = V_0 \left(\frac{B'_0}{B_0} P + 1 \right)^{-1/B'_0} \quad (2)$$

where the V_0 parameter was kept fixed for each of the α - and α' - polymorphs because of the limited pressure window available for these structures.

All of the RuBr₃ polymorphs show low bulk moduli B_0 of less than 20 GPa at ambient pressure and a strong tendency to hardening upon compression, with the B'_0

TABLE I. Parameters of the equation of state derived from the experimental pressure-dependent unit-cell volume, $V(P)$, and from total energies $E(V)$ calculated by DFT. The parameters E_0 and V_0 stand for the equilibrium energy (calculated with respect to the most stable polymorph, β -RuBr₃) and volume, respectively, whereas B_0 is the bulk modulus at ambient pressure and B'_0 is pressure derivative of the bulk modulus.

	E_0 (eV/f.u.)	V_0 (Å ³ /f.u.)	B_0 (GPa)	B'_0
<i>α-RuBr₃</i>				
Experiment		103.7	19.8(6)	6.1(7)
DFT	0.110(1)	104.7(1)	18.6(3)	6.6(2)
<i>α'-RuBr₃</i>				
Experiment		101.5	17.7(3)	6.3(2)
DFT	0.143(1)	101.5(1)	19.3(5)	7.8(3)
<i>β-RuBr₃</i>				
Experiment		107.4(2)	14.0(5)	6.5(2)
DFT	0	107.6(1)	17.7(3)	5.7(1)

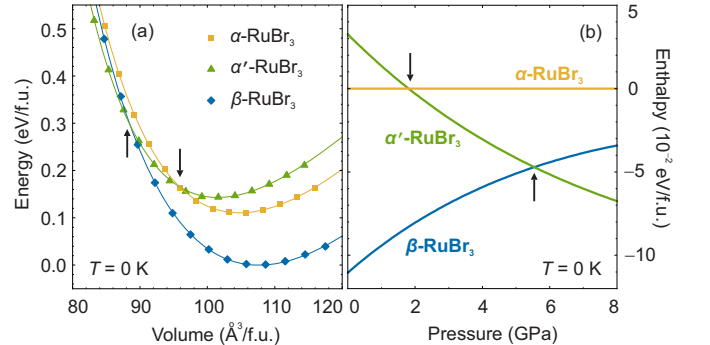


FIG. 3. (a) Total energies of the RuBr₃ polymorphs as a function of volume given relative to the energy minimum of the most stable polymorph, β -RuBr₃. The lines are the fits with Eq. (1). (b) Pressure dependence of the enthalpies given relative to α -RuBr₃. The arrows show the $\alpha - \alpha'$ and $\beta - \alpha'$ transitions

values well exceeding the typical range of $B'_0 = 4 - 5$. This elastic behavior is characteristic of van der Waals solids. For example, the $R\bar{3}$ polymorph of BiI_3 features $B_0 = 11.7(4)$ GPa and $B'_0 = 8.1(3)$ [36].

Our data suggest that β -polymorph should be the thermodynamically stable form of RuBr_3 at ambient pressure (Fig. 3a). Only at lower volume does it become less stable than α' - RuBr_3 . The magnetic α -polymorph is never the lowest-energy phase. Its formation becomes possible only because the $\alpha' - \beta$ transformation is kinetically hindered, so that α' - RuBr_3 can be quenched and gives way to the α -polymorph at lower pressures. It is worth noting that the α' - and β -polymorphs show very similar bulk moduli, as seen from their almost parallel $V(P)$ curves (Fig. 2a). The lower volume of the α' -polymorph renders it more stable under pressure.

Transition pressures are quantified by enthalpies calculated using pressure-dependent volumes extracted from the equation of state. Fig. 3b shows that the $\alpha - \alpha'$ transition should take place at 1.7 GPa in a good agreement with the zero-temperature value of 1.3 GPa determined from the magnetization measurements (Sec. III C). The XRD data show the transition at the higher pressure of 2.5 GPa at room temperature. This shift of the transition pressure with temperature should be caused by the phonon contribution to the free energy, which was not included in our DFT calculation.

The $\beta - \alpha'$ transition is expected around 5.5 GPa, but it could not be observed in our room-temperature XRD experiment because this transformation involves a major structural reorganization from chains into layers of the RuBr_6 octahedra. We argue that such a transformation must be kinetically hindered and requires elevated temperatures to be completed.

C. Magnetic properties

Fig. 4 shows the dc magnetic susceptibility M/H as a function of temperature measured under various pressures in two separate runs. In both runs, the susceptibility displays a maximum around 55 K at low pressures, followed by a kink at around $T_N \simeq 35$ K that signals the formation of long-range antiferromagnetic order [12]. Both features are rather robust against small pressures. By tracking the position of the kink at T_N , we find that the ordering temperature of α - RuBr_3 weakly increases under pressure with the slope of $dT_N/dP \simeq 1.8$ K/GPa. This increase is comparable to the changes observed in other Kitaev magnets, such as α - Li_2IrO_3 with $dT_N/dP \simeq 1.5$ K/GPa [18]. Remarkably, α - RuCl_3 shows an opposite trend, the reduction in T_N upon compression with $dT_N/dP \simeq -13.6$ K/GPa [37].

Above 1.3 GPa, the broad susceptibility maximum disappears, while a more asymmetric feature appears at T_d and rapidly shifts toward higher temperatures with increasing pressure. Concurrently, the susceptibility decreases and even becomes diamagnetic above 1.6 GPa be-

cause of the enhanced background signal as the gasket is compressed. The feature at T_d is accompanied by a temperature hysteresis and indicates a first-order phase transition that can be assigned to the $\alpha - \alpha'$ structural phase transition observed by XRD. First-order nature of this transition is further corroborated by an intermediate region with the phase coexistence. Both T_N and T_d can be observed between 1.3 and 1.5 GPa, as shown in Fig. 4b.

Our magnetization data confirm paramagnetic nature of α' - RuBr_3 , as expected from its dimerized structure. The critical pressure of the $\alpha - \alpha'$ transition is strongly temperature-dependent, similar to other Kitaev magnets [16–18, 22]. This temperature dependence is rooted in the different phonon spectra of the two polymorphs. Phonon contribution additionally stabilizes the magnetic, nondimerized phase at elevated temperatures [18], thus shifting the $\alpha - \alpha'$ transition toward higher pressures. Indeed, at room temperature this transition occurs at 2.5 GPa, as seen from our XRD data.

D. Electronic structure

Fig. 5 compares electronic structures of three RuBr_3 polymorphs calculated on the DFT+SO level. All of them show the broad valence band predominantly formed by the Br $4p$ states below -1 eV followed by two distinct complexes of the Ru t_{2g} and e_g bands, with the former lying near the Fermi level in agreement with the $4d^5$ electronic configuration of Ru^{3+} .

In contrast to the iridates [4], Ru^{3+} compounds do not show a clear splitting of the t_{2g} bands into the $j_{\text{eff}} = \frac{3}{2}$ and $j_{\text{eff}} = \frac{1}{2}$ states expected in the relativistic case, because the spin-orbit coupling constant of $\lambda \simeq 0.15$ eV for Ru^{3+} is relatively small compared to the band width. Nevertheless, the difference between the α - and α' -polymorphs is clearly visible in the vicinity of the Fermi level. The DOS of α' - RuBr_3 features a pseudogap at the Fermi level formed as a result of transforming Ru t_{2g} states into molecular orbitals driven by the short Ru–Ru bonds (dimers). Therefore, α - RuBr_3 would become insulating only upon adding electronic correlations that split the t_{2g} bands, whereas the band gap opening in α' - RuBr_3 is almost completed by the formation of the Ru–Ru bonds. We can thus classify α - RuBr_3 as Mott insulator, whereas α' - RuBr_3 is proximate to a band insulator.

These assignments are corroborated by the DFT+ U +SO calculations with $U_d = 2$ eV and $J_H = 0.26$ eV [38] that produce the magnetic insulating state for α - RuBr_3 with the band gap of 0.6 eV and Ru magnetic moment of $1.07 \mu_B$ comprising almost equal spin and orbital contributions of about $0.55 \mu_B$ each. Experimentally, insulating behavior of α - RuBr_3 has been reported at ambient pressure [39].

Another consequence of the Ru–Ru dimer formation is the broadening of the e_g bands as a result of the enhanced Ru–Br interactions. On the level of ligand-field theory

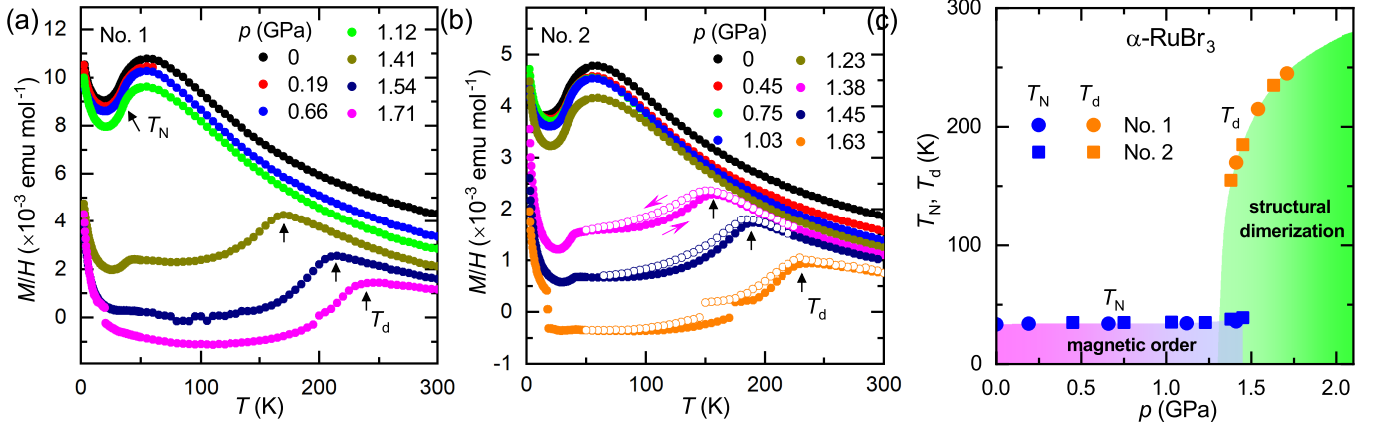


FIG. 4. Temperature-dependent dc magnetic susceptibility M/H for α -RuBr₃ measured at various pressures from 2 K to 300 K in the 1 T magnetic field for (a) run No. 1 and (b) No. 2. The labels T_N and T_d denote the antiferromagnetic ordering transition and the structural dimerization temperature, respectively. Solid symbols were collected upon warming, while open symbols were measured upon cooling. Small discontinuities in the 1.63 GPa data are due to the signal crossing zero. (c) Temperature-pressure phase diagram of α -RuBr₃.

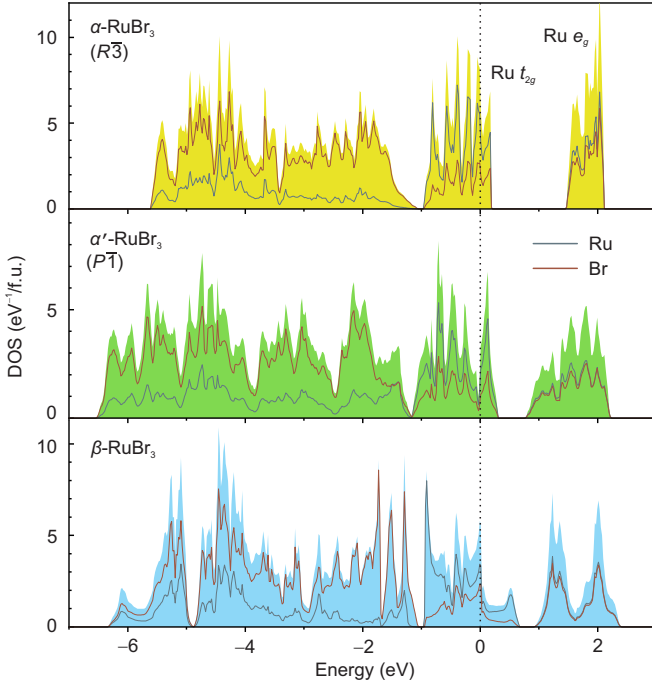


FIG. 5. Density of states (DOS) calculated on the DFT+SO level for the RuBr₃ polymorphs without spin polarization. The Fermi level is at zero energy. Ambient-pressure crystal structures of α - and β -polymorphs are used. In the case of α' -RuBr₃, we used the optimized crystal structure at 1.9 GPa, right above the transition.

for a RuBr₆ octahedron, these e_g bands can be thought as the Ru-Br antibonding states, whereas the respective bonding states occur at the bottom of the valence band, around -5 eV. Such bonding states extend to lower energies when the metal-metal bonds are formed. All these features are similar across the α' - and β -polymorphs despite their different crystal structures.

E. Magnetic interactions

To analyze pressure evolution of magnetism within the nondimerized honeycomb phase, we model α -RuBr₃ using the extended Kitaev $J - K - \Gamma - \Gamma'$ Hamiltonian for nearest-neighbor couplings in the honeycomb plane [40],

$$\mathcal{H} = \sum_{\langle ij \rangle} \left(J_{ij} \mathbf{S}_i \mathbf{S}_j + K S_i^\gamma S_j^\gamma + \Gamma (S_i^\alpha S_j^\beta + S_i^\beta S_j^\alpha) + \Gamma' (S_i^\alpha S_j^\gamma + S_i^\gamma S_j^\alpha + S_i^\beta S_j^\gamma + S_i^\gamma S_j^\beta) \right), \quad (3)$$

which is augmented by interactions beyond nearest neighbors as well as interlayer couplings. The interaction parameters are obtained from the superexchange model of Refs. [33, 34] using the on-site Coulomb repulsion $U_d = 2$ eV, Hund's coupling $J_H = 0.26$ eV, and spin-orbit coupling constant $\lambda = 0.15$ eV [38]. We find that α -RuBr₃ is dominated by the $K < 0$ and $\Gamma > 0$ terms, whereas J and Γ' are both negative and less than 1 meV in magnitude (Fig. 6b). The main sub-leading term is the third-neighbor in-plane interaction J_3 followed by the shortest interlayer coupling J_\perp , which is perpendicular to the honeycomb layers. At ambient pressure, α -RuBr₃ features $K \simeq -5$ meV, $\Gamma \simeq 2.5$ meV, and $J_3 \simeq 1.0$ meV, which is comparable to the results of the earlier *ab initio* study [41] and remarkably similar to the parameter regime established for α -RuCl₃ [42–45]. This parameter regime places α -RuBr₃ into the region of collinear zigzag order in agreement with the experiment [12].

Pressure reduces the Ru-Br-Ru bond angle and enhances Γ , while reducing the absolute value of K . Concurrently, J slightly increases in magnitude (and remains negative), whereas Γ' slightly decreases (Fig. 6b). Such changes are consistent with the expected evolution of the couplings on reducing the bond angle [34]. They also mirror pressure evolution of the exchange couplings in α -RuCl₃, although much larger changes were proposed in

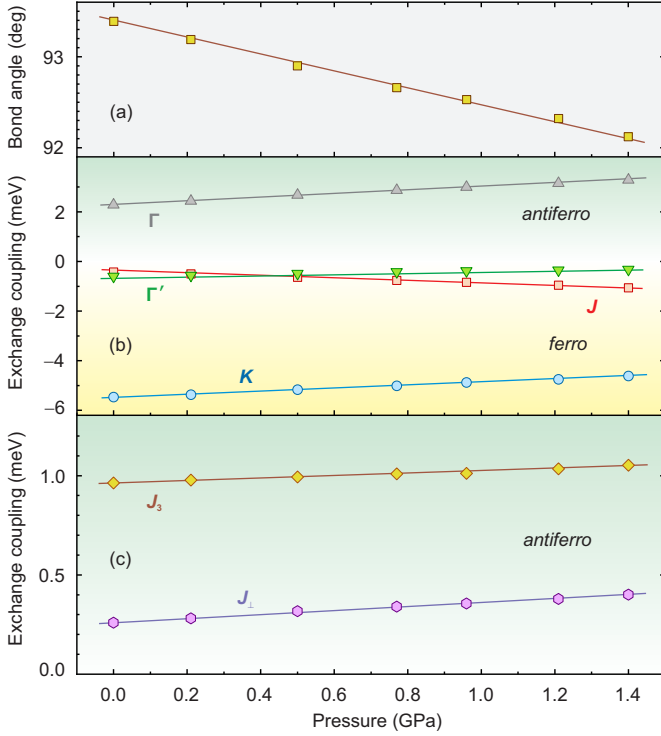


FIG. 6. Pressure dependence of the Ru-Br-Ru bond angle (a) and exchange couplings (b,c) for α -RuBr₃. The lines are guides for the eye.

that case [37].

The dissimilar trends in Γ and K suggest that the overall energy scale gauged by $\sqrt{J^2 + K^2 + \Gamma^2 + \Gamma'^2}$ remains almost constant, 5.94 meV at 0 GPa vs. 5.77 meV at 1.4 GPa, so it can't be the reason for the increase in T_N under pressure. Using our exchange parameters, we place α -RuBr₃ onto phase diagrams of the extended Kitaev model, the quantum phase diagram obtained for $\Gamma' = 0$ [46] and classical phase diagram with the small Γ' [47]. In both cases, RuBr₃ straddles the boundary between the ferromagnetic and zigzag states while moving away from the Kitaev point because $|K|/\Gamma$ decreases.

The increasing T_N indicates an additional stabilization of the zigzag state under pressure, which could hardly be explained by the nearest-neighbor couplings alone. The zigzag order in α -RuBr₃ and other Kitaev magnets is further stabilized by J_3 [34]. This coupling is enhanced under pressure and increases by 7% at 1.4 GPa. The interlayer coupling J_\perp increases by 40% in the same pressure window. Both trends could serve to explain the experimentally observed 7% increase in T_N between 0 and 1.4 GPa. It remains unclear why T_N increases in α -RuBr₃ but decreases in α -RuCl₃ under pressure [37]. In both compounds, the nearest-neighbor exchange couplings evolve in a very similar way. Therefore, the terms beyond nearest neighbors, J_3 and J_\perp , will most likely determine the T_N value, but the evolution of these terms in α -RuCl₃ has not been reported and remains an inter-

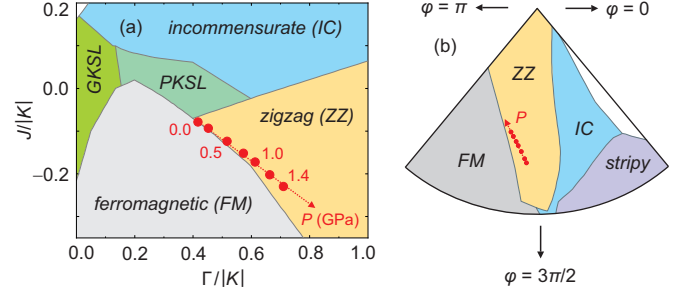


FIG. 7. Pressure evolution of RuBr₃ with respect to competing phases of the extended Kitaev model. (a) Quantum phase diagram of the $J - K - \Gamma$ model ($\Gamma' = 0$) [46]. GKSL and PKSL are the generic and proximate Kitaev spin liquids, respectively. (b) Classical phase diagram of the $J - K - \Gamma$ model with $\Gamma'/A = -0.05$ [47]. The diagram is drawn in polar coordinates: $J/A = \sin \theta \cos \varphi$, $K/A = \sin \theta \sin \varphi$, $\Gamma/A = \cos \theta$ where $0 < \theta < \pi/2$ gauges the distance from the center of the circle and $A = \sqrt{J^2 + K^2 + \Gamma^2}$ is the overall energy scale.

esting topic for future investigation.

IV. DISCUSSION AND SUMMARY

Our data show that the Kitaev candidate α -RuBr₃ is energetically less favorable than the β -polymorph and thus metastable at ambient pressure. However, the $\alpha \rightarrow \beta$ transformation is kinetically hindered and does not occur at room temperature. This allows the quenching of the honeycomb polymorph. Its synthesis from β -RuBr₃ requires not only high pressures but also elevated temperatures. The compression of β -RuBr₃ at room temperature does not suffice, as our experiments have shown, even though the transition is thermodynamically favored above 5.5 GPa.

The higher stability of β -RuBr₃ at ambient pressure is likely rooted in the large size of bromine and the more sparse nature of the chain-like structure, as opposed to the layered one. Likewise, the more compact nature of the layered structure renders it thermodynamically stable under pressure. Synthesis of α -RuBr₃ involves the $\beta \rightarrow \alpha'$ transformation, followed by the conversion of the α' -polymorph into α -RuBr₃ upon release of pressure. Only at this last step does the material become magnetic, whereas the first step of the α -RuBr₃ synthesis, the transformation between the chain and layered polymorphs, occurs between the two structures that both contain Ru dimers. These higher stability of these polymorphs can be traced back to the enhanced Ru-Br bonding. The magnetic α -polymorph exists in a narrow pressure window up to 1.3 GPa at 0 K and up to 2.5 GPa at 295 K. This pressure window is nevertheless much broader than in α -RuCl₃ where dimerization sets in already at 0.1 GPa at low temperatures [37].

From the magnetism perspective, pressure tunes α -RuBr₃ away from the Kitaev limit because it reduces

the Ru–Br–Ru bond angles and enhances Γ while reducing $|K|$. These changes do not visibly affect the position of α -RuBr₃ with respect to the boundary between the ferromagnetic and collinear zigzag states, so pressure evolution of T_N is more likely to be affected by the third-neighbor in-plane and interplane couplings, both increasing under pressure. Bringing α -RuBr₃ closer to the Kitaev limit would require an expansion of the structure via negative pressure. Partial iodine substitution [39, 48] and strain tuning [49] may be useful in this context.

In summary, we have studied pressure evolution of the different RuBr₃ polymorphs and revealed magnetism collapse of the honeycomb α -polymorph, the structural sibling of the renowned α -RuCl₃. Our data suggest that the pressure-induced transformation from the chain structure into the honeycomb structure takes place between the two paramagnetic phases that both contain the Ru–Ru dimers. These dimers disappear upon release of pressure, giving way to the magnetic α -polymorph, which is metastable. The pressure window of this magnetic α -

RuBr₃ is somewhat larger than in the chloride, but it does not exceed that of the Kitaev iridates. Hydrostatic pressure tunes α -RuBr₃ away from the Kitaev limit, so that expansion rather than contraction of the structure would be necessary in order to enhance Kitaev interactions in this material.

Experimental and computational data associated with this manuscript are available from Refs. [50, 51].

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* These authors contributed equally

† altsirlin@gmail.com

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Supplemental Material

Magnetic vs. nonmagnetic polymorphs of RuBr_3 under pressure

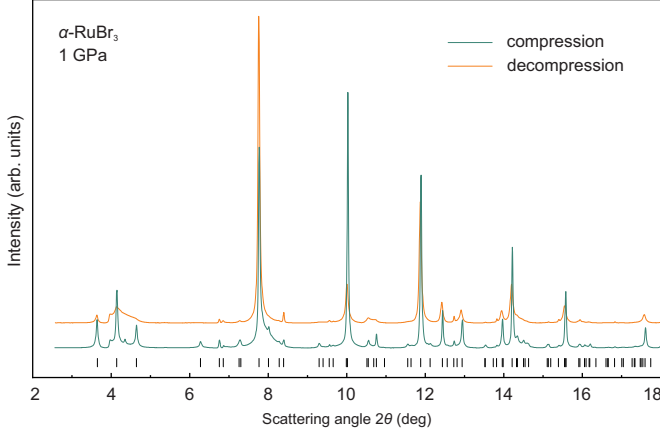


FIG. S1. XRD patterns of $\alpha\text{-RuBr}_3$ collected around 1 GPa upon compression and decompression. The patterns are offset for clarity.

Fig. S1 shows the XRD patterns of $\alpha\text{-RuBr}_3$ collected upon compression and decompression. They demonstrate reversibility of the $\alpha - \alpha'$ transition. However, the crystallinity of the sample is notably reduced after pressure cycling.

Fig. S2 displays pressure-dependent lattice parameters of $\beta\text{-RuBr}_3$. At room temperature and ambient pressure, β -polymorph has an orthorhombic ($Pmmn$) crystal structure with the ordered arrangement of the Ru–Ru dimers [11, 13]. Above 384 K, it reversibly transforms into a hexagonal ($P6_3/mcm$) structure where similar chains with the alternation of short and long Ru–Ru distances become disordered relative to each other [11]. Low crystallinity of our $\beta\text{-RuBr}_3$ sample did not allow us to resolve the orthorhombic superstructure. Therefore, we analyzed the XRD data for $\beta\text{-RuBr}_3$ using hexagonal symmetry and determined two lattice parameters, a and c , that reflect the interchain and intrachain distances, respectively.

Pressure evolution of $\beta\text{-RuBr}_3$ reveals a small hysteresis between 5 and 10 GPa indicating a possible phase transition associated with the shortening of the c parameter. Details of this transition could not be resolved in the present experiment, owing to the low crystallinity of the $\beta\text{-RuBr}_3$ sample. However, it is worth noting that the transition happens in the pressure range where the c -parameter approaches 5.6 Å, which is about twice the

value of the Ru–Ru intradimer distance at ambient pressure (2.71 – 2.73 Å [11, 13]). Therefore, it is plausible that the discontinuity in the c lattice parameter of $\beta\text{-RuBr}_3$ corresponds to a transformation within the dimerized Ru chains, indicating for example the loss of dimerization, but further experiments performed on a sample

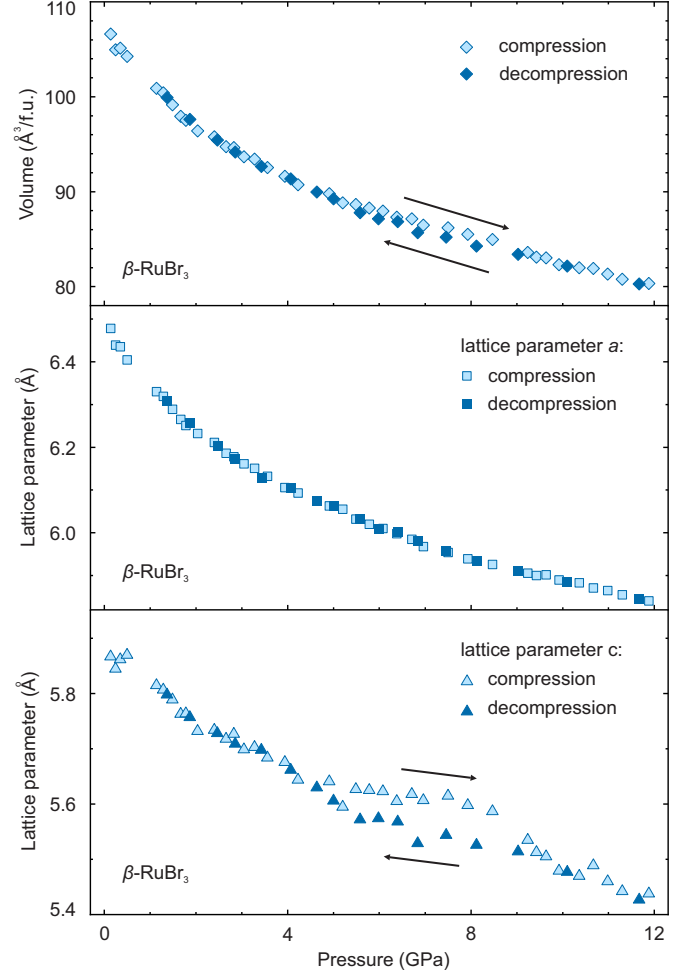


FIG. S2. Pressure dependence of the unit-cell volume and lattice parameters for $\beta\text{-RuBr}_3$.

with higher crystallinity would be necessary in order to determine the Ru–Ru distances as a function of pressure and verify this conjecture.